

AP Environmental Science Exam Prep Session: Energy Resources and Consumption

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AP Environmental Science Topic Outline

- V. Energy Resources and Consumption (10–15%)
 - A. Energy Concepts
(Energy forms; power; units; conversions; Laws of Thermodynamics)
 - B. Energy Consumption
 - 1. History
(Industrial Revolution; exponential growth; energy crisis)
 - 2. Present global energy use
 - 3. Future energy needs
 - C. Fossil Fuel Resources and Use
(Formation of coal, oil, and natural gas; extraction/purification methods; world reserves and global demand; synfuels; environmental advantages/disadvantages of sources)
 - D. Nuclear Energy
(Nuclear fission process; nuclear fuel; electricity production; nuclear reactor types; environmental advantages/disadvantages; safety issues; radiation and human health; radioactive wastes; nuclear fusion)
 - E. Hydroelectric Power
(Dams; flood control; salmon; silting; other impacts)
 - F. Energy Conservation
(Energy efficiency; CAFE standards; hybrid electric vehicles; mass transit)
 - G. Renewable Energy
(Solar energy; solar electricity; hydrogen fuel cells; biomass; wind energy; small-scale hydroelectric; ocean waves and tidal energy; geothermal; environmental advantages/disadvantages)

Important Legislation

Low-Level Radioactive Policy Act: all states must have facilities to handle low-level radioactive wastes.

Nuclear Waste Policy Act: US government must develop a high level nuclear waste site by 2015 (see Yucca Mountain).

Important Terms

First Law of Thermodynamics: energy is neither created nor destroyed, but may be converted from one form to another.

Second Law of Thermodynamics: when energy is changed from one form to another, some useful energy is always degraded into lower quality energy (usually heat).

Ionizing radiation: radiation w/enough energy to free electrons from atoms forming ions, may cause cancer (ex. gamma, X-rays, UV).

High Quality Energy: organized & concentrated, can perform useful work (ex. fossil fuels & nuclear).

Low Quality Energy: disorganized, dispersed (ex. heat in ocean or air/wind, solar).

Natural radioactive decay: unstable radioisotopes decay releasing gamma rays, alpha & beta particles (ex. Radon).

Half-life: the time it takes for 1/2 of the mass of a radioisotope to decay. A radioactive isotope must be stored for approximately 10 half-lives until it decays to a safe level.

Nuclear Fission: nuclei of isotopes split apart when struck by neutrons.

Nuclear Fusion: 2 isotopes of light elements (H) forced together at high temperatures until they fuse to form a heavier nucleus; happens in the Sun; very difficult to accomplish on Earth; prohibitively expensive.

Electricity Generation: steam, from water boiled by fossil fuels or nuclear energy, or falling water is used to turn a generator.

Petroleum (Crude Oil) Formation: microscopic aquatic organisms in sediments converted by heat & pressure into a mixture of hydrocarbons.

Petroleum Pros: cheap, easily transported, high-quality energy.

Petroleum Cons: reserves depleted soon, pollution during drilling, transport and refining, land subsidence, burning oil produces CO₂.

Coal Formation: prehistoric plants buried un-decomposed in oxygen-depleted water of swamps/bogs converted by heat and pressure.

Ranks of Coal: peat, lignite, bituminous coal, anthracite coal (from least to best)

Nuclear Reactor: consists of a core, control rods, moderator, steam generator, turbine, containment building.

Alternative Energy Sources: wind, solar, waves, biomass, geothermal, fuel cells